Battery Materials

# Rechargeable Battery Reagents





# **Class of Reagents**

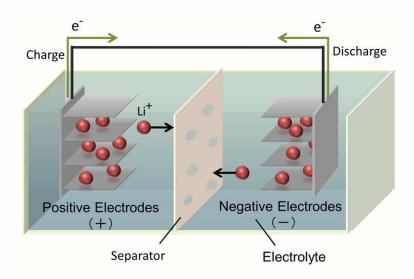
- Lithium-Ion Battery
- Toshima Manufacturing, Ltd.,
   Materials of Lithium-Ion Batteries
- SHINSHU VOLTA CO., LTD., Materials of Lithium-Ion Batteries
- Magnesium-Ion Battery
- Sodium-Ion Battery
- Ionic Liquids etc.

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# Lithium-Ion Battery

Lithium-ion batteries are used in a wide range of products due to their high energy density and voltage. Because of the development of electric vehicles and mobile devices, the demand for high-capacity, high-voltage, and safe secondary batteries is growing. FUJIFILM Wako deals with various materials that constitute lithium-ion batteries.



#### Positive Electrodes(+)

(Discharge) (Charge)  $\text{Li}_1\text{-xCoO}_2 + \text{xLi}^+ + \text{xe}^- \rightleftarrows \text{LiCoO}_2$ 

#### **Negative Electrodes(—)**

(Discharge) (Charge)  $Lix6C \rightleftharpoons 6C + xLi^+ + xe^-$ 

#### Mechanism of the LIB system

#### Positive Electrodes

Manufacturer	Code No.	Product Name	Formula	Storage Condition CAS RN <sup>®</sup>	Pack Size
FUJIFILM Wako Pure Chemical	125-03501	Lithium Mahahata	Li MoO	Keep at RT	100g
	127-03505	Lithium Molybdate	Li <sub>2</sub> MoO <sub>4</sub>	13568-40-6	500g

#### Negative Electrodes

Manufacturer	Code No.	Product Name	Formula	Storage Condition CAS RN <sup>®</sup>	Pack Size
FUJIFILM	107.00001	Lithium Chat	1:	Keep at RT	10~
Wako Pure Chemical	127-06001	6001 Lithium, Shot	Li	7439-93-2	10g
FUJIFILM Wako Pure	208-01602	Tin(IV) Oxide (98.0+%)	SnO₂	Keep at RT	25g
Chemical	202-01605		31102	18282-10-5	500g
FUJIFILM Wako Pure	191-05582	22 Silican Baudan 00 00/	Si	Keep at RT	25g
Chemical	191-05562	Silicon, Powder, 99.9%	) Ji	7440-21-3	
FUJIFILM Wako Pure	198-05612	Silican Manavida, 00 09/	0:0	Keep at RT	25.0
Chemical	190-05012	Silicon Monoxide, 99.9%	SiO	10097-28-6	25g
FUJIFILM 190-09072 Shake Pure Chemical 192-09071	Silipan Diavida 00 0%	0:0	Keep at RT	25g	
	192-09071	Silicon Dioxide, 99.9%	SiO <sub>2</sub>	7631-86-9	100g

# Electrolytes

Manufacturer	Code No.	Product Name	Formula	Melting point (ref.)	Storage Condition CAS RN®	Pack Size
FUJIFILM	122-06632	Lithium Bis(fluorosulfonyl)imide	0. Li <sup>†</sup> 0	4.4000	Keep at RT	25g
Wako Pure Chemical	124-06631	[Synonyms : LiFSI]	F O O F	140°C	171611-11-3	100g
FUJIFILM Wako Pure	120-06851	Lithium (Fluorosulfonyl) (trifluoromethanesulfonyl)imide	O Li <sup>†</sup>	100°C	Keep at RT	. 5g
Chemical	120-00031	[Synonyms : LiFTFSI]	F O O CF3	100 0	192998-62-2	39
FUJIFILM Wako Pure	129-06642	Lithium Bis(trifluoromethanesulfonyl)imide	O Li <sup>†</sup> O O S CF <sub>3</sub>	234-	Keep at RT	25g
Chemical	121-06641	[Synonyms : LiTFSI]	F <sub>3</sub> C S CF <sub>3</sub>	238°C	90076-65-6	100g
FUJIFILM	400 00700	Lithium Trifluoromethanesulfonate	0, 0	423°C	Keep at RT	25g
Wako Pure Chemical	128-03792	Littilium miliuoromethanesunonate	O, O F <sub>3</sub> C S OLi	423 C	33454-82-9	23g
FUJIFILM	327-61481	Lithium 1,1,2,2,3,3,4,4,4-Nonafluoro- 1-butanesulfonate	0, .0	370-	Keep at RT	5g
Wako Chemical	325-61482		$O_{5}O$ $F_{9}C_{4}$ OLi	380°C	131651-65-5	25g
FUJIFILM	123-01542	Listerium A a state	0	280-	Keep at RT	25g
Wako Pure Chemical	127-01545	Lithium Acetate	H₃C OLi	285°C	546-89-4	500g
FUJIFILM	121-05921	likhirum llavafirananha amhata¥1		165-	Keep at RT	10g
Wako Pure Chemical	127-05923	Lithium Hexafluorophosphate*1	LiPF <sub>6</sub>	175°C	21324-40-3	50g
FUJIFILM	123-06042	I :4b:: D		0000	Keep at RT	25g
Wako Pure Chemical	125-06041	Lithium Perchlorate*1	LiClO <sub>4</sub>	236°C	7791-03-9	100g
FUJIFILM	128-06031	Listering Tetrafficants and #1		00000	Keep at 2-10 °C	5g
Chemical	/ako Pure Lithium Tetrafluoro Chemical 126-06032		LiBF <sub>4</sub>	293°C	14283-07-9	25g
FUJIFILM	404 00704	Lithium	O Li <sup>+</sup> O F <sub>3</sub> C-S-CF <sub>3</sub>		Keep at RT	1
Wako Pure Chemical	121-06761	06761 Tris(trifluoromethanesulfonyl)methide [Synonyms : LiTFSM]		-	132404-42-3	1g

<sup>※1</sup> Battery Grade

# Solvents of the Electrolyte Solution

Manufacturer	Code No.	Product Name	Formula	Storage Condition CAS RN <sup>®</sup>	Pack Size
FUJIFILM	169-25201	Propylene Carbonate	CH₃	Keep at RT	100mL
Wako Pure Chemical	161-25205	【Synonyms : PC】*1		108-32-7	500mL
FUJIFILM Wako Pure	057-08491	Ethylene Carbonate		Keep below 25 °C	100g
Chemical	059-08495	【Synonyms : EC】*1	O O	96-49-1	500g
FUJIFILM Wako Pure	063-06711	Fluoro Ethylene Carbonate	F	Keep at 2-10 °C	5g
Chemical	061-06712	【Synonyms : FEC】*1		114435-02-8	25g
FUJIFILM Wako Pure	044-31931	Dimethyl Carbonate	H₃CO OCH₃	Keep at RT	100mL
Chemical	046-31935	【Synonyms : DMC】*1	0	616-38-6	500mL
FUJIFILM Wako Pure	047-31921	Diethyl Carbonate	H <sub>3</sub> C O O CH <sub>3</sub>	Keep at RT	100mL
Chemical	049-31925	【Synonyms: DEC】*1	0	105-58-8	500mL

<sup>※1 =</sup> Battery Grade

# $\bullet {\sf Specification} \\$

			Specification		
	【PC】 _CH3	[EC]	[FEC]	[DMC]	[DEC]
				H₃CO_OCH₃ 0	H <sub>3</sub> C O O CH <sub>3</sub>
Purity (cGC)	≧98.0%	≧98.0%	≧99.5%	≧98.0%	≧98.0%
Water Content	≦20ppm	≦50ppm	≦200ppm	≦20ppm	≦20ppm
Acid (as H <sub>2</sub> CO <sub>3</sub> )	-	-	-	≦0.1%	≦0.2%
Chloride	≦5ppm	≦5ppm	-	≦5ppm	≦5ppm
Са	≦1.0ppm	≦1.0ppm	-	≦1.0ppm	≦1.0ppm
Fe	≦1.0ppm	≦1.0ppm	-	≦1.0ppm	≦1.0ppm
К	≦1.0ppm	≦1.0ppm	-	≦1.0ppm	≦1.0ppm
Na	≦1.0ppm	≦1.0ppm	-	≦1.0ppm	≦1.0ppm

# Cross-linked Polyacrylic Acids, "CLPA Series"

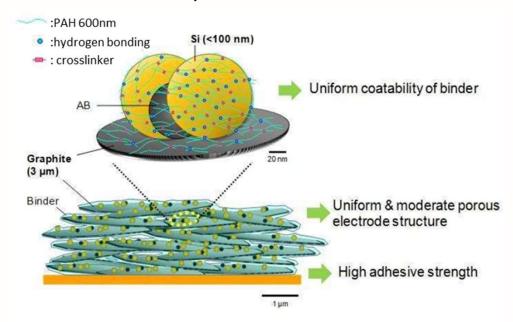
The CLPA series is a water-based binder developed for the high capacity Si negative electrode for next generation lithium-ion secondary batteries.

The expansion and shrinkage of Si during charge and discharge causes isolation of Li-Si, inhibiting capacity degradation.

**CLPA** structure

#### **Features**

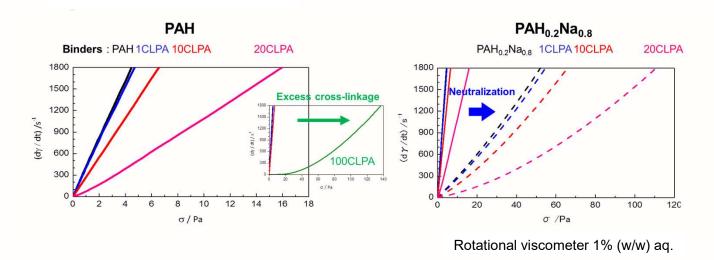
- The chemically cross-linked polyacrylic acid shows increased viscosity and thixotropy\* when it is neutralized.
- Active materials are contained and the electrode destruction is prevented by the physical and chemical crosslinking, so that the cycle characteristic can be improved.
  - \*Thixotropy is the property that the viscosity of a material decreases with time when the material is under constant shear stress and viscosity increases with time when the stress is removed.



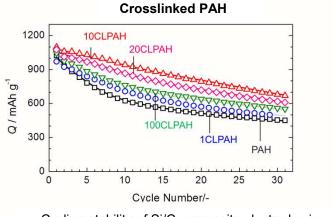
Manufacturer	Code No.	Product Name	Storage Condition	CAS RN®	Pack Size
FUJIFILM Wako Chemical	355-41122	PAH (non-crosslinked type)	Keep at RT	9003-01-4	25g
FUJIFILM Wako Chemical	352-41132	1 CLPAH	Keep at RT	9003-01-4	25g
FUJIFILM Wako Chemical	359-41142	10 CLPAH	Keep at RT	9003-01-4	25g
FUJIFILM Wako Chemical	356-41152	20 CLPAH	Keep at RT	9003-01-4	25g
FUJIFILM Wako Chemical	353-41162	100 CLPAH	Keep at RT	9003-01-4	25g

#### Physical and Electrochemical Properties of CLPA

#### Viscosity of Binder Solution

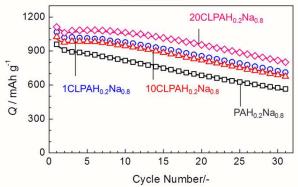


#### Capacity retention; Effect of Covalent Crosslinkage & Partial Neutralization [a]



Cycling stability of Si/C composite electrodes is Improved by using the moderately crosslinked PAH binder

#### Crosslinked PAH<sub>0.2</sub>Na<sub>0.8</sub> (partially neutralized)



Cycling stability of Si/C composite electrodes is further improved by "neutralization" of binders.

[a] Komaba, S. et al.: J. Electrochem. Soc, 162., A2245 (2015).

# Toshima Manufacturing Co., Ltd., Materials of Lithium-Ion Batteries



FUJIFILM Wako provides lithium-ion battery materials manufactured by Toshima Manufacturing Co., Ltd. A company engaged in the development and marketing of positive electrode materials used in lithium-ion batteries and solid electrolytes, which have been attracting public attention, can provide such materials with high purity. Materials and shapes not listed in the lineups can be provided. Please contact if the product you want is not included in the list.

#### Positive Electrodes

Code No.	Manufacturer Code No.	Product Name	Purity	Appearance	Size	Storage Condition CAS RN <sup>®</sup>	Pack Size	
						Keep at RT		
381-04661	LiLBPW01	LiCoO <sub>2</sub>	3N*²	Powder	0.5 <b>~</b> 1µm	12190-79-3	100g	
000 04004	L'II DDIAGO	1.31.0	201			Keep at RT	400	
389-04601	LiLBPW03	LiNiO <sub>2</sub>	3N	Powder	5∼15µm	12031-65-1	100g	
000 04004	I : I DDIMOE	1.5	011	-		Keep at RT	400	
382-04831	LiLBPW05	LiFeO <sub>2</sub>	3N	Powder	5 <b>~</b> 15µm	12022-46-7	100g	
005 04004	LUBBIAGE		201			Keep at RT	400	
385-04681	LiLBPW07	Li <sub>2</sub> MnO <sub>3</sub>	3N	Powder	5 <b>~</b> 15µm	-	100g	
000 04044	1:1 0014100	1.34	011			Keep at RT	400	
386-04611	LiLBPW09	LiMn <sub>2</sub> O <sub>4</sub>	3N	Powder	5∼15µm	12057-17-9	100g	
						Keep at RT	100g	
385-04701	LiLBPW13	LiNi <sub>0.5</sub> Mn <sub>1.5</sub> O <sub>4</sub>	3N	Powder	30∼50 <i>µ</i> m	-		
						Keep at RT	100g	
-	LiLBPW16	LiCo <sub>1/3</sub> Ni <sub>1/3</sub> Mn <sub>1/3</sub> O <sub>2</sub>	3N	Powder	5∼15 <i>µ</i> m	346417-97-8		
						Keep at RT		
383-04621	LiLBPW18	LiFePO <sub>4</sub>	3N	Powder	5 <b>~</b> 15µm	15365-14-7	100g	
						Keep at RT		
380-04631	LiLBPW20	LiCoPO <sub>4</sub>	3N	Powder	1∼10 <i>µ</i> m	13824-63-0	100g	
						Keep at RT		
388-04671	LiLBPW22	LiNiPO <sub>4</sub>	3N	Powder	1∼10 <i>µ</i> m	13977-83-8	100g	
						Keep at RT		
382-04691	LiLBPW24	LiMnPO <sub>4</sub>	3N	Powder	1∼10 <i>µ</i> m	13826-59-0	100g	
000 40004		Liv (DO)	011	D 1	- 4-	Keep at RT	400	
382-19221	-	Li <sub>3</sub> V <sub>2</sub> (PO <sub>4</sub> ) <sub>3</sub>	3N	Powder	5 <b>~</b> 15µm	-	100g	
200 04044		LiCa Ni Mar C	211	Daviden	4 40 +2	Keep at RT	400	
389-04841	-	LiCo <sub>0.2</sub> Ni <sub>0.4</sub> Mn <sub>0.4</sub> O <sub>2</sub>	3N	Powder	1∼10 <i>µ</i> m*³	-	100g	

**<sup>%</sup>**2 3N=99.9%

**<sup>%</sup>**3 Estimation

Code No.	Manufacturer Code No.	Product Name	Purity	Appearance	Size	Storage Condition CAS RN <sup>®</sup>	Pack Size		
						Keep at RT			
-	-	NCM523 (LiNi <sub>0.5</sub> Co <sub>0.2</sub> Mn <sub>0.3</sub> O <sub>2</sub> )	3N*²	Powder	5 <b>~</b> 15µm	-	100g		
			011544		5∼15µm	Keep at RT	100g		
-	-	NCM622(LiNi <sub>0.6</sub> Co <sub>0.2</sub> Mn <sub>0.2</sub> O <sub>2</sub> )	2N5* <sup>4</sup>	Powder		-			
						Keep at RT	100g		
-	-	NCM811(LiNi <sub>0.8</sub> Co <sub>0.1</sub> Mn <sub>0.1</sub> O <sub>2</sub> )	2N5	Powder	5 <b>~</b> 15µm	-			
								Keep at RT	
-	-	1.15C+LiFePO₄	3N	Powder	1~10µm	-	100g		
				_		Keep at RT	100		
-	-	2C+LiMn <sub>0.6</sub> Fe <sub>0.4</sub> PO <sub>4</sub>	3N	Powder	1∼10µm	-	100g		

¾4 2N5=99.5%

# Negative Electrodes

Code No.	Manufacturer Code No.	Product Name	Purity	Appearance	Size	Storage Condition CAS RN <sup>®</sup>	Pack Size
385-04561	Lil RDW56	LiLBPW56 Li <sub>4</sub> Ti <sub>5</sub> O <sub>12</sub>	3N	Powder	4 40	Keep at RT	100g
303-04301	LILDI VV30			1 Owder	1∼10 <i>µ</i> m	-	
		- C (Spherical graphite : C-Dm10) Powder		Keep at RT			
-	-			Powder	9 <b>~</b> 11µm	-	100g
		C (Cubovicel growthite C Due15)	4N* <sup>5</sup>	Davidan		Keep at RT	
-	-	C (Spherical graphite : C-Dm15)	TIN	Powder	14∼16µm	-	100g
		C (Spherical graphite - C Dm22)		Powder		Keep at RT	
-	-	C (Spherical graphite : C-Dm22)			21~23µm	-	100g

**<sup>※</sup>**5 4N=99.99%

#### All solid Electrolytes for all solid-state Batteries

All solid-state lithium-ion batteries that use no organic solvent are the subject of active research to develop an incombustible and safer next-generation battery. Toshima Manufacturing Co., Ltd. offers various materials, such as NASICON or perovskite type materials, representative solid electrolyte materials and garnet type materials with excellent oxidation/reduction stability and large potential window.

Code No.	Manufacturer Code No.	Product Name	Purity	Appearance	Size	Storage Condition CAS RN <sup>®</sup>	Pack Size
381-13152	LiLBPW26	Li <sub>6.25</sub> La <sub>3</sub> Zr <sub>2</sub> Al <sub>0.25</sub> O <sub>12</sub> (cubic)	3N	Powder	5 <b>~</b> 10µm	Keep at RT	25g
301-13132	LILDI VVZO	Li <sub>6.25</sub> La <sub>3</sub> Li <sub>2</sub> Ai <sub>0.25</sub> O <sub>12</sub> (cubic)	514	1 OWGCI	υ τομπ	-	239
388-13162	LiLBPW28	Li <sub>6.6</sub> La <sub>3</sub> Zr <sub>1.6</sub> Ta <sub>0.4</sub> O <sub>12</sub> (cubic)	3N	Powder	5 <b>~</b> 10µm	Keep at RT	25g
300-13102	LILDI VVZO	Li <sub>6.6</sub> La <sub>3</sub> Li <sub>1.6</sub> i a <sub>0.4</sub> O <sub>12</sub> (cubic)	514	1 OWGCI	σετομπ	-	209
389-04802	LiLBPW32	Li <sub>7</sub> La <sub>3</sub> Zr <sub>2</sub> O <sub>12</sub> (tetra)	3N	Powder	5~15µm	Keep at RT	25g
303-04002	LILDI WJZ	Li <sub>7</sub> La <sub>3</sub> Zi <sub>2</sub> O <sub>12</sub> (tetta)	JIN	1 Owder	5 15μπ	-	239
387-04722	LiLBPW36	Li <sub>0.33</sub> La <sub>0.55</sub> TiO <sub>3</sub> (tetra)	3N	Powder	5 <b>~</b> 10µm	Keep at RT	- 25g
307-04722	LILDI WOO	Li <sub>0.33</sub> La <sub>0.55</sub> 11O <sub>3</sub> (tetia)	JIN	i owdei		-	
388-04752	LiLBPW42	Li <sub>1.5</sub> Al <sub>0.5</sub> Ge <sub>1.5</sub> P <sub>3</sub> O <sub>12</sub>	3N	Powder	5~15µm	Keep at RT	25g
300-04732	LILDI VV42	(amorphous)	JIN	i owdei	5 15μπ	-	239
381-04742	LiLBPW44	Li <sub>1.5</sub> Al <sub>0.5</sub> Ge <sub>1.5</sub> P <sub>3</sub> O <sub>12</sub>	3N	Powder	5 <b>~</b> 15µm	Keep at RT	25g
301-04742	LILDF VV44	(rhomb)	JIN	rowdei	<i>5</i> ~ 1 <i>5</i> μπ	-	239
385-13172	LiLBPW46	Li AL TI DO	3N	Powder	5 <b>~</b> 15µm	Keep at RT	25g
303-13172	LILDF VV40	Li <sub>1.3</sub> Al <sub>0.3</sub> Ti <sub>1.7</sub> P <sub>3</sub> O <sub>12</sub>	JIN	rowdei	<i>5</i> ~ 1 <i>5</i> μπ	-	239
387-04641	LiLBPW48	Li DO	3N	Powder	1 Euro	Keep at RT	100g
307-04041	LILDF VV40	Li <sub>3</sub> PO <sub>4</sub>	JIN	rowdei	1∼5µm	10377-52-3	Toog
385-04821	1 LiLBPW54 Li <sub>3</sub> BO <sub>3</sub> 3N Powder 10~30μm	Keep at RT	100=				
303-04021	LILDI- VV J4	Li <sub>3</sub> BO <sub>3</sub>	JIN	i owaei	10 30μπ	13774-56-6	100g

#### Others

Code No.	Manufacturer Code No.	Product Name	Purity	Appearance	Size	Storage Condition CAS RN <sup>®</sup>	Pack Size	
386-04655	NaNBPW01	Na₃PO₄	3N	Powder	5~15µm	5~15um	Keep at RT	500g
300-04033 Nainb	Naindi Woi	Na <sub>3</sub> i O <sub>4</sub>	Va <sub>3</sub> i O <sub>4</sub>	1 OWGC1		7601-54-9	Joog	
206 04054	NaNDDW02	No 7r Si DO	2N* <sup>6</sup>	2N*6 Powder	5~15µm	Keep at RT	100g	
386-04851	NaNBPW03	Na <sub>3</sub> Zr <sub>2</sub> Si <sub>2</sub> PO <sub>12</sub>				-		

**<sup>%</sup>**6 2N5 =99%

#### Introduction of the Barrel Sputtering Technique

The barrel sputtering is a coating technique for powders and the sputter deposition is used to deposit films primarily on substrates (flat surfaces). Feeding the base material powder into a polyhedral barrel and rotating it to form a film enables a surface to be coated homogeneously and efficiently with the powder. This powder coating used for various purposes such as catalysts, raw materials of cosmetics, decoration, and raw materials of ceramics has drawn attention in recent years as a manufacturing technique for lithium-ion batteries.

Through a joint research collaboration with Professor Abe of the University of Toyama, Toshima Manufacturing Co., Ltd. is engaged in the contract manufacturing of powder coating films (sputter/surface modification).

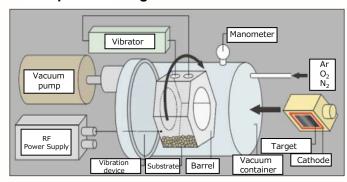
#### **OIntended Purposes**

- Positive Electrodes of Li Batteries
- Catalyst-related Applications
- Decoration

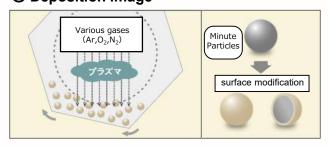
#### O Equipment appearance



#### O Composition Image



#### O Deposition Image



#### <Example of a Past Successful Result>

NCM111 (LiNi<sub>0.33</sub>Co<sub>0.33</sub>Mn<sub>0.33</sub>O<sub>2</sub>) LiNbO<sub>3</sub> Deposition on Substrates

#### **Before Deposition**



LiNbO<sub>3</sub> on Substrates

Substrates: LiNi<sub>0-33</sub>Co<sub>0-33</sub>Mn<sub>0-33</sub>O<sub>2</sub> Powder

# After Deposition SEM TEM LiNbO3 fillm NCM111 Nb SEM-EDX Nb Nb NcM111

#### SHINSHU VOLTA CO., LTD., Materials of Lithium-Ion Batteries





SHINSHU VOLTA CO.,LTD. established in May 2021 (Representative: Takeshi Hashimoto, MEIJO NANO CARBON Co., Ltd.) is a venture company originating from Shinshu University. The company develops lithium-ion batteries based on the research results that were obtained in the development of lithium-ion battery materials, which has been promoted by the leading researcher, Professor Nobuyuki Zettsu in Faculty of Engineering/Director of ELab² Center, Next-generation Research Center Cluster, Interdisciplinary Cluster for Cutting Edge Research (ICCER) of Shinshu University, and were licensed in part to MEIJO NANO CARBON Co., Ltd.

#### Monocrystal"LiCoO₂(LCO)"

#### "The world's first commercialization of a battery material grown by the flux method"

Using the flux method (a crystal growth method of recrystallizing inorganic material crystals using a fused salt as a solvent) to grow monocrystalline LCO particles, which are widely used for the positive electrodes of lithium ion batteries to be installed to small consumer electronics and drones, has made it possible to achieve a smaller average size (conventional product) and more excellent 10C rate property (can be fully discharged in 6 minutes) than those of existing products, and significantly improve the capacity decrease in charge and discharge cycles at 10C rate compared with the conventional products. By using the flux method, which is the advantage of Shinshu University, monocrystalline particles with a structure widely covered with a {104} face advantageous to lithium-ion diffusion can be obtained, whereas the LCO particles of the conventional products are amorphous and polycrystalline.

Toc rate operation

140

140

140

140

Flux growth LCO

Flux growth LCO

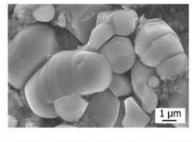
Cycle

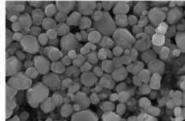
Quick
discharging

Li<sub>1-x</sub>CoO<sub>2</sub>

Li<sub>1-x</sub>CoO<sub>2</sub>

conventional product : high-output LCO





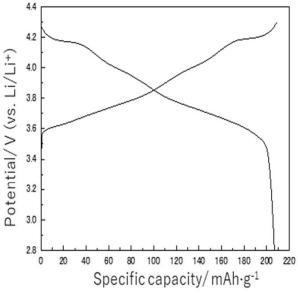
Shindai Crystal LCO

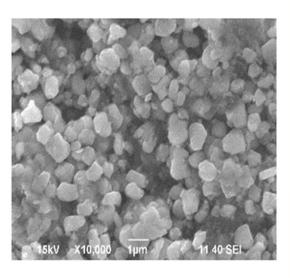
Data resulting from the discharge test of the cell with the monocrystalline LCO grown by the flux method and FE-SEM images of the electrode surfaces after the cycle test. Shindai Crystal hardly deteriorates even after the high-rate discharge test at 10C rate.

Code No.	Product Name	Formula	Storage Condition CAS RN®	Pack Size
636-53211	Lithium Cahalt Ovida	1:000	Keep at RT	100
	Lithium Cobalt Oxide	LiCoO₂	12190-79-3	10g

# LiNi<sub>0.9</sub>Co<sub>0.05</sub>Mn<sub>0.05</sub>O<sub>2</sub> (NCM91) Electrode with Bimodal Average Particle Size Distribution

As a next-generation positive electrode for high-capacity lithium-ion batteries used in the electric vehicles, a high nickel layered tertiary positive electrode material whose basic composition is LiNi<sub>x</sub>Co<sub>y</sub>Mn<sub>1-x-y</sub>O<sub>2</sub> has been gathering attention. An increased nickel content results in an increase in specific capacity of the positive electrode material but it makes the manufacturing process far more difficult. Through the combination of the precursor manufacturing technique of Tsukishima Kikai Co., Ltd. and the synthesis technique of Shinshu University, a NCM91 positive electrode with bimodal average particle size distribution has been developed. This increases the packing density of the NCM91 particles in the electrode and makes the resulting potential distribution uniform, leading to an extended battery life. A specific capacity of at least 200 mAh/g, which is higher than those of the positive electrode materials used in the conventional electric vehicles, can also be obtained.



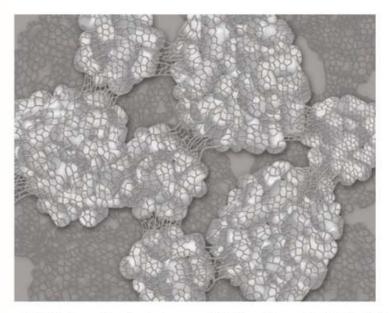


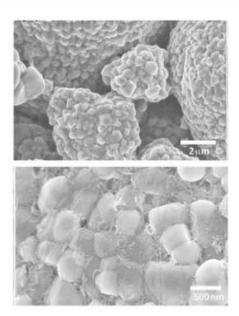
Charge/discharge curves and FE-SEM image of the  $LiNi_{0.9}Co_{0.05}Mn_{0.05}O_2$  (NCM91) positive electrode with bimodal average particle size distribution. A specific capacity of about 210 mAh/g can be obtained in the cut-off voltage range of 2.8-4.3 V (vs.Li/Li<sup>+</sup>).

Code No.	Product Name	Formula	Storage Condition CAS RN <sup>®</sup>	Pack Size	
633-53221	Lithium Nickel Manganese Cobalt Oxide	LiNi Co Mp O	Keep at RT	10g	
033-33221	[NCM91]	$\text{LiNi}_{0.9}\text{Co}_{0.05}\text{Mn}_{0.05}\text{O}_{2}$	346417-97-8	iug	

#### Carbon Nanotube Dispersion Liquid [high output]

The world's first conductive binder has been developed by the creation of a hybrid of single-walled and multi-walled carbon nanotubes. This brings a success in significantly reducing the amount of a conductive aid or binder to be used, which does not contribute to the battery capacity, and enables a high-energy density electrode with high concentrations of active materials. The carbon nanotube forms an electronic conduction network that spreads three-dimensionally and uniformly in the electrode, making it possible to significantly reduce conventional electrode resistance. It has been shown clear that the reticulately spreading electronic conduction network serves as a binder to bind the adjacent active material particles existing in the electrode.



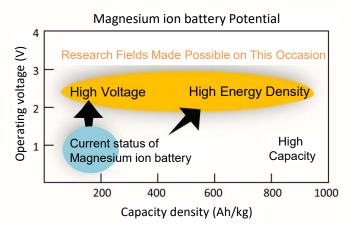


Diagrammatic illustration of low-modulus carbon nanotube binder and FE-SEM images of an NCM electrode with high concentrations of active space: materials (NCM: CNT = 99.5: 0.5 (wt%)). The company provides electrodes with input-output characteristics, high energy density, and cycle characteristic.

Code No.	Product Name	Formula	Storage Condition CAS RN <sup>®</sup>	Pack Size
632-53235	Carbon Nanotube Dispersion Liquid	С	Keep at RT	500ml
032-33233	[high output]	C	-	500mL

#### **Magnesium-Ion Battery**

As a successor to lithium-ion batteries, for which there is concern about resource depletion, magnesium batteries have been gathering attention. Although magnesium is a promising material for secondary batteries because it is more abundant and safer than lithium, magnesium batteries have room for improvement compared to lithium-ion batteries in terms of energy density and cycle characteristics. Therefore, the development of advanced electrodes and electrolytes is expected for magnesium batteries.



#### Positive Electrodes

Manufacturer	Code No.	Product Name	Formula	Storage Condition CAS RN®	Pack Size
FUJIFILM Wako Pure Chemical	222-00122	Vanadium(V) Oxide	V <sub>2</sub> O <sub>5</sub>	Keep at RT	25g
	226-00125	variauluiii(v) Oxide	V <sub>2</sub> O <sub>5</sub>	1314-62-1	500g

#### Negative Electrodes

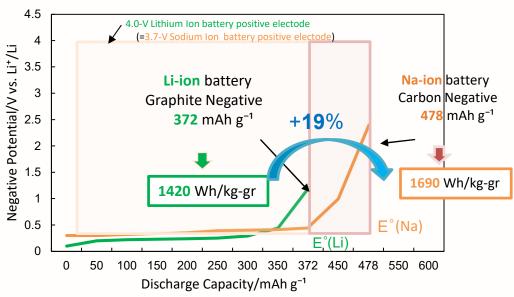
Manufacturer	Code No.	Product Name	Formula	Storage Condition CAS RN®	Pack Size
FUJIFILM Wako	201 15201	Tin, Powder, -45µm, 99.5%	5	Keep at RT	100~
Pure Chemical	201-15301	ΠΠ, Powder, -45μΠ, 99.5%	Sn	7440-31-5	100g

#### Electrolytes

Manufacturer	Code No.	Product Name	Formula	Storage Condition CAS RN <sup>®</sup>	Pack Size
FUJIFILM Wako Pure	133-07981	Magnesium Perchlorate	Mg(ClO <sub>4</sub> ) <sub>2</sub>	Protect from light	50g
Chemical	139-07983	300~850µm(20~48mesh)	Ivig(OiO <sub>4</sub> ) <sub>2</sub>	10034-81-8	400g
FUJIFILM	195-05703	- Sulfolane	0,,0	Keep at RT	25mL
Wako Pure Chemical	199-05706	Sullolarie		126-33-0	500mL

#### **Sodium-Ion Battery**

Sodium secondary batteries are made from raw materials that are more abundant than those for lithium-ion batteries, so they can be produced at a lower cost than lithium-ion batteries. Although sodium secondary batteries were considered to have lower energy density than lithium-ion batteries in the past, it is expected that with a newly developed negative electrode material, they will achieve energy density equivalent to that of lithium-ion batteries<sup>[b]</sup>.



[b] Azusa. Kamiyama. et al.: Angew. Chem. Int. Ed., 60, 5114 (2021).

#### Negative Electrodes

Manufacturer	Code No.	Product Name	Formula	Storage Condition CAS RN <sup>®</sup>	Pack Size
FUJIFILM Wako Pure Chemical	201-15301	Tin, Powder, -45µm, 99.5%	Sn	Keep at RT 7440-31-5	100g
FUJIFILM Wako	204-09862	Tin(II) Oxide, 99.9%	SnO	Keep at RT	25g
Pure Chemical	204-09002	Till(11) Oxide, 55.570	5110	21651-19-4	239

#### Electrolytes

Manufacturer	Code No.	Product Name	Formula	Storage Condition CAS RN <sup>®</sup>	Pack Size
FUJIFILM Wako	195-01462	Sodium Tetrafluoroborate	NoDE	Keep at RT	25g
Pure Chemical	199-01465	Sodium retranuoroporate	NaBF₄	13755-29-8	100g
FUJIFILM Wako	198-09252	Sodium Perchlorate	NaCIO	Keep at RT	25g
Pure Chemical	192-09255	Sodium Perchiorate	NaClO <sub>4</sub>	7601-89-0	500g
FUJIFILM Wako	100 10001	Sodium Bis(fluorosulfonyl)imide	Q Na <sup>+</sup>	Keep at RT	Fa
Pure Chemical	190-18801	[Synonyms : NaFSI]	F O O F	100669-96-3	5g
FUJIFILM Wako	107 10011	Sodium (Fluorosulfonyl)	Q Na <sup>+</sup> Q	Keep at RT	Ea
Pure Chemical	197-10011	97-18811 (trifluoromethanesulfonyl)imide [Synonyms : NaFTFSI]		1233836-95-7	5g

#### **Ionic Liquids**

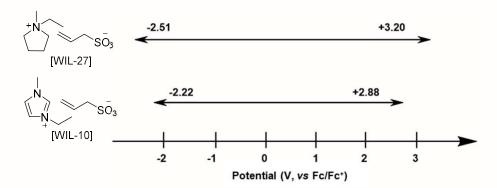
#### Battery Grade [WIL Series]

lonic liquids, which are refractory, flame-retardant, and heat-resistant and safer than organic solvents, have been receiving attention as a flame-retardant material. Research and development of ionic liquids as electrolytes and electrolyte solutions for lithium-ion secondary batteries have been widely proceeded with from the viewpoint of enhanced safety, in particular, such as preventing ignition and explosion of batteries, and they are expected to be put into practical use as a material for secondary batteries.

FUJIFILM Wako has developed the WIL series, original ionic liquids suitable for research of batteries. The series has the advantage of having a larger potential window than the existing ionic liquids.

#### What is a Potential Window?

Potential window is defined as an electrolyte range where current does not flow in the electrolyte solution, that is, a range that an oxidation-reduction reaction does not take place on the electrodes and is used as an indicator of electrochemical properties.



Code No. Product Name	Product Pack		Storage Condition	Viscosity	Melting	Conductivity	Electric Potential	
		oddol T doll	(mPa · s, 25°C)	Point (°C)	Conductivity (mS/cm, 25°C)	Area (V)	Width (V)	
052-08561	\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\	1g Keep at RT 312 -8	-83	1.2	-2.51 ~	5.71		
058-08563	WIL-27	5g	1268622-88-3	312	-03	1.2	+3.20 (Fc)	5.71
056-08581	WIL-10	1g	Keep at RT	210	-68	1.0	-2.22 ~ 5.40	5.10
052-08583		5g	1268622-81-6	210		1.0	+2.88 (Fc)	5.10

# **Ionic Liquids as Precursors**

The anion exchange method is one of the synthesis methods of ionic liquids. An ionic liquid can be prepared by the reaction of a desired cationic haloid salt with a desired anionic lithium salt. Potassium salt and sodium salt are also offered. Consider the use of the products for the synthesis of ionic liquids.

Manufacturer	Code No.	Product Name	Formula	Melting Point (ref.)	Storage Condition CAS RN®	Pack Size
FUJIFILM Wako Pure	122-06632	Lithium Bis(fluorosulfonyl)imide	O. N. O	4.40.0	Keep at RT	25g
Chemical	124-06631	[Synonyms : LiFSI]	F OOF	140°C	171611-11-3	100g
FUJIFILM Wako Pure	190-18801	Sodium Bis(fluorosulfonyl)imide	O Na <sup>†</sup>	112°C	Keep at RT	Fa
Chemical	190-10001	[Synonyms : NaFSI]	F S T S	112 C	100669-96-3	5g
FUJIFILM Wako Pure	167-28742	Potassium Bis(fluorosulfonyl)imide	0 K <sup>+</sup> 0	101°C	Keep at RT	25g
Chemical	169-28741	[Synonyms : KFSI]	F S F	101 C	14984-76-0	100g
FUJIFILM Wako Pure	120-06851	Lithium (Fluorosulfonyl) (trifluoromethanesulfonyl)imide	Q Li <sup>†</sup>	100°C	Keep at RT	5g
Chemical	120-00001	[Synonyms : LiFTFSI]	F S CF <sub>3</sub>	100 C	192998-62-2	Jy .
FUJIFILM Wako Pure	197-18811	Sodium (Fluorosulfonyl) (trifluoromethanesulfonyl)imide	Na <sup>†</sup> O N- O		Keep at RT	5g
Chemical		[Synonyms : NaFTFSI]	F S CF <sub>3</sub>	-	1233836-95-7	Jy .
FUJIFILM Wako Pure		Potassium (Fluorosulfonyl) (trifluoromethanesulfonyl)imide	0 N- 0	101°C	Keep at RT	5g
Chemical		100-20731	[Synonyms : KFTFSI]	F O O CF3	101 C	860653-59-4
FUJIFILM Wako Pure	129-06642	Lithium Bis(trifluoromethanesulfonyl)imide	Q Li <sup>†</sup> Q	234-	Keep at RT	25g
Chemical	121-06641	(Synonyms : LiTFSI)	F <sub>3</sub> C S CF <sub>3</sub>	238°C	90076-65-6	100g
FUJIFILM Wako Pure	167-28002	Potassium Bis(trifluoromethanesulfonyl)imide	0 K <sup>+</sup>	200°C	Keep at RT	25g
Chemical	169-28001	(Synonyms : KTFSI)	$F_3C$ $O$ $O$ $CF_3$	200 C	90076-67-8	100g
FUJIFILM Wako Pure	128-03792	Lithium Trifluoromethanesulfonate	0,0	423°C	Keep at RT	25g
Chemical			F <sub>3</sub> C´ OLi		33454-82-9	_
FUJIFILM Wako	327-61481	Lithium 1,1,2,2,3,3,4,4,4-Nonafluoro- 1-butanesulfonate	0,0	370-	Keep at RT	5g
Chemical	325-61482	1-butariesuiloriate	F <sub>9</sub> C <sub>4</sub> OLi	380°C	131651-65-5	25g
FUJIFILM Wako Pure	123-01542	Lithium Acetate	0	280-	Keep at RT	25g
Chemical	127-01545	Elimani / toolato	H <sub>3</sub> C OLi	285°C	546-89-4	500g
FUJIFILM	198-01072		0	>300°C	Keep at RT	25g
Wako Pure Chemical	190-01071	<u> </u>	H <sub>3</sub> C ONa			100g
Chemical	192-01075				127-09-3	500g

Manufacturer	Code No.	Product Name	Formula	Melting Point (ref.)	Storage Condition CAS RN <sup>®</sup>	Pack Size
FUJIFILM Wako Pure	166-03172	Potassium Acetate	0	292°C	Keep at RT	25g
Chemical	160-03175		H₃C CK	292°C	127-08-2	500g
FUJIFILM Wake Pure	121-05921	Lithium	LiPF <sub>6</sub>	165-	Keep at RT	10g
Wako Pure Chemical	127-05923	Hexafluorophosphate*1		175°C	21324-40-3	50g
FUJIFILM Wako Pure	128-06031	Lithium Tetrafluoroborate*1	LiBF₄	293°C	Keep at 2-10 °C	5g
Chemical	126-06032	Littliam retrailabloborate	LIDITA	233 0	14283-07-9	25g
FUJIFILM Wako Pure	195-01462	Cadirum Tatuaflu anala anata	5-	384°C	Keep at RT	25g
Chemical	199-01465	Sodium Tetrafluoroborate	NaBF <sub>4</sub>	304 C	13755-29-8	500g
FUJIFILM Wako Pure	161-16792	Potassium Tetrafluoroborate	KDE	520°C	Keep at RT	25g
Chemical	165-16795	rotassium retranuoroporate	KBF₄	530°C	14075-53-7	500g

<sup>※1</sup> Battery Grade

#### Cation Precursors

Manufacturer	Code No.	Product Name	Formula	Melting Point (ref.)	Storage Condition CAS RN®	Pack Size
FUJIFILM Wako Pure	134-12801	1-Methylimidazole	ÇH₃ ∕N	-60°C	Protect from light	100mL
Chemical	136-12805	wetrymmidazore		-00 0	616-47-7	500mL
FUJIFILM Wako Pure	133-08042	1 Mathydninoridina	ÇH₃ ∠N∖	-18°C	Keep at RT	25mL
Chemical	137-08045	1-Methylpiperidine		-10 C	626-67-5	500mL
FUJIFILM	132-06873	4 Mathadasanahalina	ÇH₃ √N√	66.6	Keep at RT	25mL
Wako Pure Chemical	136-06876	4-Methylmorpholine		-66°C	109-02-4	500mL

Manufacturer	Code No.	Product Name	Formula	Melting Point (ref.)	Storage Condition CAS RN®	Pack Size
FUJIFILM Wako Pure Chemical	164-05312	Pyridine	N	-42°C		25mL
	162-05313				Keep at RT	100mL
	166-05316				110-86-1	500mL
FUJIFILM Wako Pure Chemical	161-02603	- 4-Methylpyridine	H <sub>3</sub> C	2.4°C	Keep at RT	25mL
	165-02606				108-89-4	500mL
FUJIFILM Wako Pure Chemical	200-07723	Tributyl Phosphine	P( <i>n</i> -C <sub>4</sub> H <sub>9</sub> ) <sub>3</sub>	-	Keep at RT	25mL
	204-07726				998-40-3	500mL
FUJIFILM Wako Pure Chemical	208-03662	Trimethylammonium Chloride	H <sub>3</sub> C + - H <sub>3</sub> C -NH CI H <sub>3</sub> C	277°C	Keep at RT	25g
					593-81-7	
FUJIFILM Wako Pure Chemical	204-02363	Tributylamine	N( <i>n</i> -C <sub>4</sub> H <sub>9</sub> ) <sub>3</sub>	-70°C	Keep at RT	25mL
	208-02366				102-82-9	500mL
FUJIFILM Wako Pure Chemical	205-09272	- Trioctylamine	N( <i>n</i> -C <sub>8</sub> H <sub>17</sub> ) <sub>3</sub>		Keep at RT	25mL
	209-09275			-	1116-76-3	500mL

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